

## Chapter Test B Arrangement Of Electrons In Atoms

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the periodic table, Mendeleev based his arrangement of the elements on an element's atomic mass

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Modern Chemistry 31 Chapter Test Chapter: Arrangement of Electrons in Atoms PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_\_ 1. The principal quantum number of an electron is 4. What are the

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Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued \_\_\_\_\_ 8. What is the frequency of a photon whose energy is  $3.4 \times 10^{-19}$  J? ( $h = 6.626 \times 10^{-34}$  J-s) a.  $8.8 \times 10^{26}$  Hz b.  $5.1 \times 10^{14}$  Hz c.  $1.9 \times 10^{-15}$  Hz d.  $2.3 \times 10^{-52}$  Hz \_\_\_\_\_ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's surface.

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(b) particle nature. (d) wave nature. 6. c According to Bohr, which is the point in the figure below where electrons cannot reside? (a) point A (c) point C (b) point B (d) point D 7. c According to the quantum theory, point D in the above figure represents (a) the fixed position of an electron. (b) the farthest position from the nucleus that an electron can occupy.

4 Arrangement of Electrons in Atoms

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\_\_\_\_\_ refers to the arrangement of conditions in a study so that meaningful comparisons of the effect of the IV can be made 1. intervention 2. experimental design 3. single-subject design 4. group comparison design

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