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A)8, 2, and 3 B)7, 1, and 2 C)2, 7, and 4

D)2, 4, and 6 E)1, 5, and 7 6) Lattice energy
is _____. A)the sum of electron affinities of
the components in an ionic solid B)the
energy required to convert a mole of ionic
solid into its constituent ions in the gas

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phase C)the sum of ionization energies of the components in an ionic solid

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Chapter 8: Reactions of Alkenes -

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Flow Chart for selection of test for gas 8.

Test For Anions ANIONS TEST 1

Carbonate ion, CO_3^{2-} - Pour 3 cm³ of aqueous carbonate solution into a test tube - Add 2 cm³ hydrochloric acid, HCl - Bubble the gas produced through limewater - Lime water turned milky. 2 Chloride ion, Cl^-

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Pour about 2 cm³ of aqueous chloride solution into test tube

Chapter 8: Salts - Pasti Skor

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activity series, then metal X. loses electrons less readily than does metal Y. A chemical equation is balanced when the same number of each kind of atom appears in the reactants and in the products.

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Chapter 8 Review Multiple Choice Identify the choice that best completes the statement or answers the question. ___ A___ 1. In a chemical reaction a. the mass of the reactants equals the mass of the products. b. the mass of the products is greater than the

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mass of reactants. c. the number of atoms in... <https://studyres.com/doc/2433046/honors-chemistry-chapter-8-test-review-answer-key>.

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