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(Experiment 13.1) $\text{HCl (aq) + H}_2\text{O (l)} \rightleftharpoons \text{H}_3\text{O}^+ \text{ (aq) + Cl}^- \text{ (aq)}$ (Experiment 13.2) $\text{HCl (aq)} \rightleftharpoons \text{H}^+ \text{ (aq) + Cl}^- \text{ (aq)}$ Strong bases are substances that ionize completely in water producing hydroxide ions. The ionization of sodium hydroxide, a strong base, is shown as the simplified form in Experiment 13.3.

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