

Chemquest 22 Covalent Bonding Answer Key

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Ionic vs. Molecular

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Covalent Bond and its Types Lewis Structure Chemistry for Class 11 in Hindi
Draw Covalent Bonding Molecules Chemquest 22 Covalent Bonding Answer

A covalent bond is the type of bond between two nonmetals. Covalent bonds are formed by neutral atoms that share electrons rather than by charged ions. When a compound is formed by sharing electrons, the compound is called a molecule or molecular compound. It is important to note that ionic compounds are not called molecules.

Information : Terminology

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The number of electrons that should appear in the bonding picture for CO₃ is 22. The number of electrons that appear in the picture for CO₃²⁻ is 24. Offer an explanation for why CO₃²⁻ has 24 electrons instead of 22. (Where did the extra two electrons come from?)

ChemQuest 23 : Drawing Covalent Compounds

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