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Determining Numbers of Ions Use with Chapter 7, Section 7.3

1. Write a simple word equation that illustrates what must be true of total positive charge and total negative charge in an ionic compound.
2. How many potassium ions (group 1) would be related to balance the charge of each of the following in a compound?

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Chemistry: Matter and Change Chapter 7 ...

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The exponents are added. $(2.00 \times 10^4) = 2.00 \times 10^3 \times 10^1$
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MATH SKILLS TRANSPARENCY WORKSHEET Use with Chapter 1, Section 11.1 MATH SKILLS TRANSPARENCY WORKSHEET 16 Section 11.2 Determine the mole ratios for each of the following balanced chemical equations.

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1. $2\text{C}(s) + \text{O}_2(g) \rightarrow 2\text{CO}(g)$ 1 mol O_2 2 mol C 2 mol CO / 2 mol C 2 mol C / 1 mol O_2 2 2 mol CO / 1 mol O_2 2 mol C / 2 mol CO 1 mol O_2 / 2 mol CO 2 2. $\text{WO}_3(s) + 3\text{H}_2(g) \rightarrow \text{W}(s) + 3\text{H}_2\text{O}(l)$

MATH SKILLS TRANSPARENCY MASTER 16

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